



## Extrinsic semiconductors-Fermi level

why because the carrier conc. at Room Temp. & conductivity are very low example Si:

$$n_i = 10^{10} \text{ cm}^{-3} \quad n_i^2 = 3 \times 10^6 \text{ cm}^{-1}$$

$$NP = n_i^2 = \text{constant at given Temp}$$

so we can not increase p and n at same time

n-Type - P, As, Sb      Group 5

one extra electron

p-Type - B, Al, Ga, In = 3

So far we only talk about Si, what about other semiconductor and doping in them? if we look at the Germanium (Ge), Germanium lies in the same group of silicon, which means all the elements that we used in silicon could also be used for Ge. So we can use group 5 elements like phosphorous or arsenic has n-type dopant, and could also use group 3 elements like boron as a p-type dopant. we can calculate the ionization energies for these dopants, could use hydrogen model

only different the effective mass  $\frac{m_e^*}{Ge} \neq \frac{m_e^*}{Si}$  and

$$\begin{aligned} Ge \quad E_v &= 16 \\ Si \quad E_v &= 11.9 \end{aligned}$$

n-Type dopants of Ionization Energy



in mev $\Rightarrow$	P	As	Sb	$E_g(Im.)$
Si	4.5	5.4	3.9	1.10 eV
Ge	12	12.7	9.6	0.67 eV